

M.Sc.(Chem.) Semester - 1 (CBCS) Examination
OCT/NOV-2024 (AS PER SYLLABUS YEAR JUNE-2023)
Analytical Chemistry (Core (New-23))

Time: 2:30 Hours

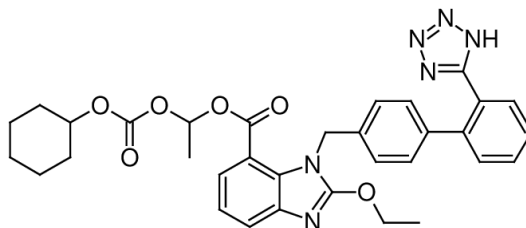
Marks: 70

Instructions:

1. All questions are compulsory.
2. Figures to the right indicate marks.

Q.1 (a) Answer the following question**(04)**

- (1) How would you prepare 100 mL of a 10 ppm solution of aspirin in methanol?
- (2) Write empirical formula and calculate molecular weight:

**Q.1 (b) Answer any two questions out of three.****(10)**

- (1) Define Molarity and explain preparation of following buffers in DM water:
 (A) 500 mL of 0.1 M hydrochloric acid and (B) 1000 mL 20 mM sulfuric acid.
- (2) Define Analytical Chemistry and differentiate between classical methods and modern Instrumental methods.
- (3) Explain mole concept and find the number of moles present in 196 gm of H_2SO_4 .

Q.2 (a) Answer the following question**(04)**

- (1) Explain refraction and refractive index with figure.

Q.2 (b) Answer any two questions out of three.**(10)**

- (1) Write a note on immersion refractometer with figure.
- (2) Give the process for calibration of polarimeter
- (3) Discuss instrumentation of polarimeter with diagram.

Q.3 (a) Answer the following question**(04)**

- (1) Explain Mohr's method and its limitation.

Q.3 (b) Answer any two questions out of three.**(10)**

- (1) Write a note on Fajan's method for estimation of chloride.
- (2) Explain types of Ligands with example used in complexometric titration.
- (3) Give classification of volumetric titration and enlist the basic requirements of volumetric methods.

Q4 (A) Answer the following questions**(04)**

- (1) Write twelve principles of green chemistry.

Q4 (B) Answer any two questions out of three**(10)**

- (1) Explain atom economy with suitable example.
- (2) Discuss synthesis of Ibuprofen in classic route and green route.
- (3) Write a note on global carbon cycling and resource depletion

Q.5 (a) Answer the following question**(04)**

- (1) Explain glass transition temperature in brief.
- (2) Name various thermocouples used in DSC.

Q.5 (b) Answer any two questions out of three.**(10)**

- (1) Write a note on factors influencing TGA and DTA analysis.
- (2) Give the principle and working of DSC apparatus.
- (3) Give principle of TGA and discuss TGA thermogram of $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$.
